CHAPTER 9

WASTE STABILIZATION POND

WASTE STABILIZATION PONDS

- Waste stabilization ponds are shallow man-made basins. It can be round, square or rectangular.
- Stabilization ponds have been employed for treatment of wastewater for over 300 years.

ADVANTAGES

Simplicity

- □ simple to construct
- simple to operate and maintain
- only unskilled labour is needed

Low Cost

cheaper than other wastewater treatment processes
 no need for expensive equipment

High Efficiency

- □ BOD removals > 90%
- Total nitrogen removals is 70-90%
- Total phosphorus removal is 30-45%
- Efficient in removing pathogens

THE PRINCIPAL REQUIREMENTS FOR WSP

- Sufficient land is available
- The soil should preferably have a coefficient of permeability less than 10⁻⁷ m/s (to avoid the need for pond lining)

TYPES OF WSP

- 1. Facultative pond
- 2. Maturation pond

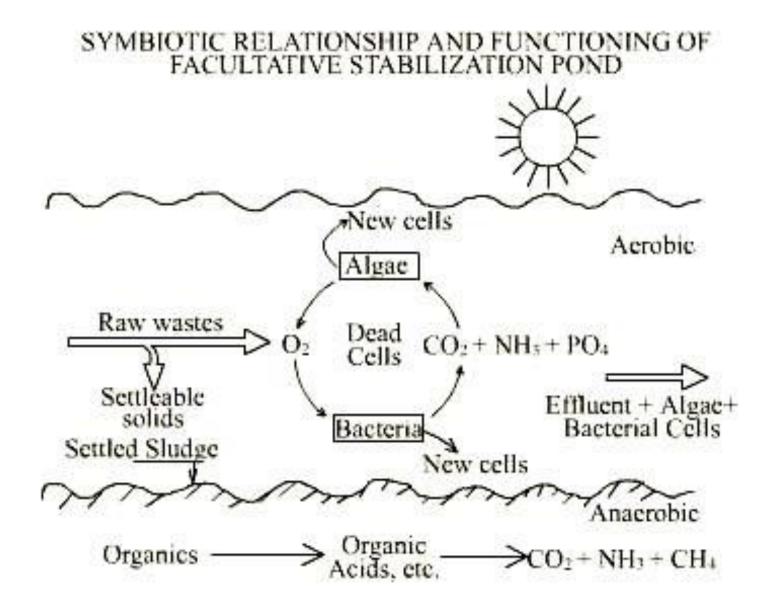
FACULTATIVE POND

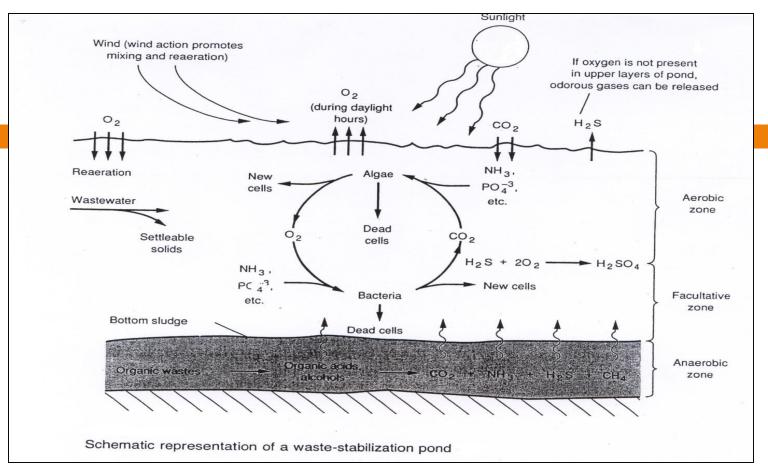
□ 1-2 m deep

□ The primary function is the removal of BOD

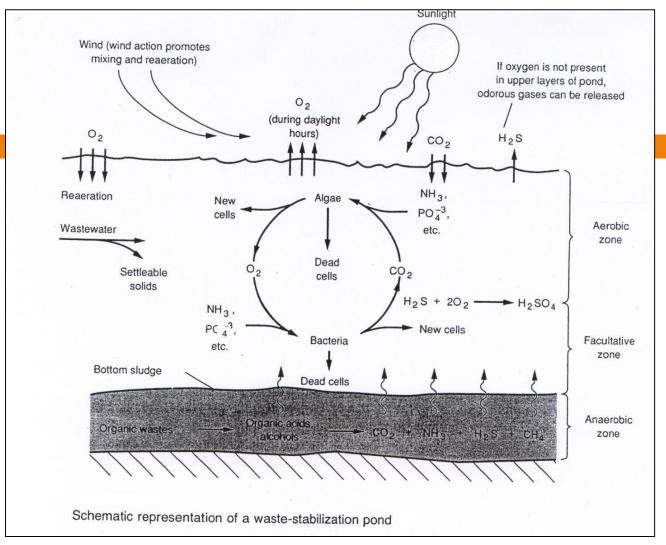
3 ZONE EXIST:

- A surface zone where aerobic bacteria and algae exist in a symbiotic relationship. The algae provide the bacteria with oxygen and the bacteria provide the algae with carbon dioxide.
- b) An anaerobic bottom zone in which accumulated solids are decomposed by anaerobic bacteria.
- c) An intermediate zone that is partly aerobic and partly anaerobic in which the decomposition of organic wastes is carried out by facultative bacteria.

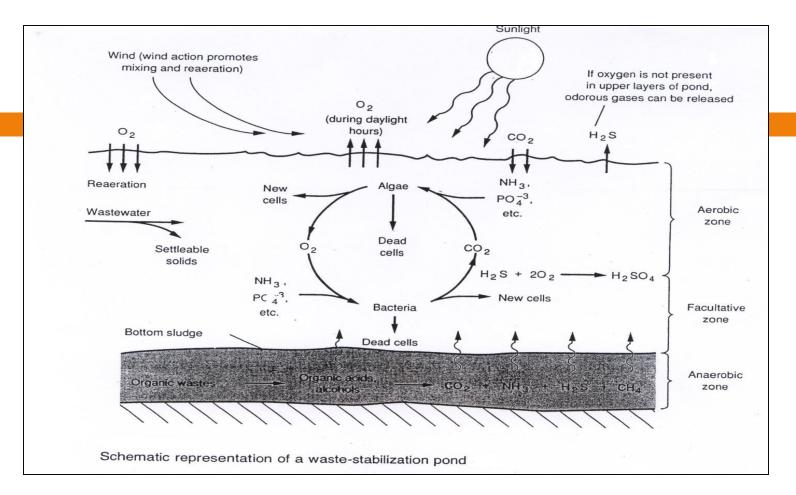




A surface zone where aerobic bacteria and algae exist in a symbiotic relationship. The algae provide the bacteria with oxygen and the bacteria provide the algae with carbon dioxide.



An anaerobic bottom zone in which accumulated solids are decomposed by anaerobic bacteria.



An intermediate zone that is partly aerobic and partly anaerobic in which the decomposition of organic wastes is carried out by facultative bacteria.

MATURATION POND

- □ 1-1.5 m deep
- Receive the effluent from a facultative pond
- Primary function is the removal of pathogens

BOD REMOVAL

In facultative ponds, BOD removal is achieved by sedimentation of settleable solids and the remaining non-settleable BOD is oxidized by heterotrophic bacteria

In maturation ponds only a small amount of BOD removal occurs

PATHOGEN REMOVAL

Bacteria

Faecal bacteria are mainly removed in facultative and especially maturation ponds

PATHOGEN REMOVAL (CONT.)

The <u>principal mechanism</u> for faecal bacteria removal are:

Time and temperature

- Faecal bacteria die-off in ponds increase with both time and temperature

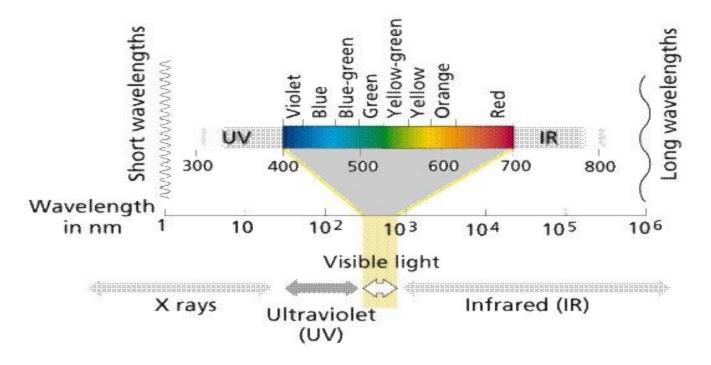
• High pH:

- Faecal bacteria (except Vibrio Cholerae) die very quickly (within minutes) at pH>9

PATHOGEN REMOVAL (CONT.)

• High light intensity

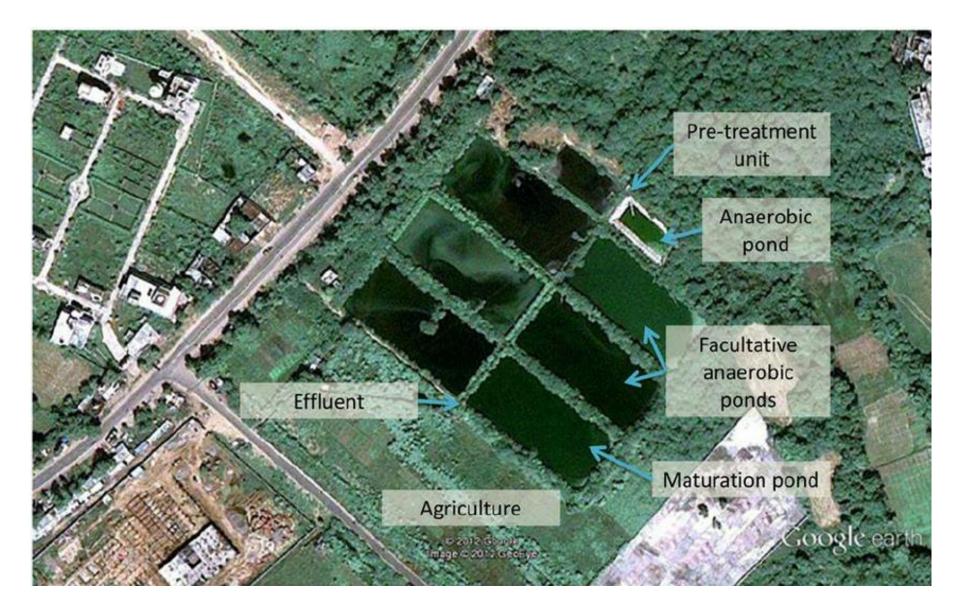
- Light of wavelength 425 – 700 nm can damage faecal bacteria



PATHOGEN REMOVAL (CONT.)

Parasites

protozoan cyst and helminth eggs are removed by sedimentation



DESIGN OF WSP

FACULTATIVE PONDS

Surface BOD loading (λ_s , kg/ha d)

$$\lambda_{\rm s} = 10 L_{\rm i} Q / A_{\rm f}$$
 (9.1)

where $A_f = facultative pond area, m^2$

The permissible BOD loading, $\lambda_{\text{s,max}}$

$\lambda_{s,max} = 350 (1.107 - 0.002T)^{T-25}$ (9.2)

Once a suitable value of λ_{c} has been selected, the pond area is calculated from equation (9.2) and its retention time (t_f, d) from: $t_f = A_f D/Q$ (9.3)Where

- D = pond depth, m
- $Q = wastewater flow, m^3/day$

MATURATION PONDS

a) Faecal Coliform Removal

Faecal coliform removal can be modeled by first order kinetics in a completely mixed reactor.

The resulting equation for a single pond is:

$$N_e = N_i / (1 + k_T t)$$
(9.4)

Where N_e = number of FC per 100 mL of effluent N_i = number of FC per 100 mL of influent k_T = first order rate constant for FC removal, per day t = retention time, day For a series of facultative and maturation ponds, equation (8.4) becomes:

$$N_e = \frac{N_i}{\left(1 + K_T t_f\right) \left(1 + K_T t_m\right)^n}$$

Where N_e and N_i now refer to the numbers of FC per 100 mL of the final effluent and raw wastewater respectively, and <u>n is the number of maturation ponds.</u>

The value of k_T is highly temperature dependent. $k_T = 2.6 (1.19)^{T-20}$ (9.6) Check the BOD effluent concentration, l_e

$$l_{e} = \frac{l_{i}}{K_{1}t + 1}$$
(9.7)
Where $l_{e} = BOD$ effluent concentration, mg/L
 $l_{i} = BOD$ influent concentration, mg/L
 $K_{1} =$ first order rate constant for BOD removal,
per day
t = retention time, day

The value of K_1 is highly temperature dependent

$$K_{1(T)} = K_{1(20)} 1.05^{T-20}$$
(9.8)

where $K_1 @ 20^{\circ}C = 0.3$ per day

For n ponds in series, BOD effluent can be calculated as follows

$$l_{e} = \frac{l_{i}}{\left(1 + K_{1}t_{f}\right)\left(1 + K_{1}t_{m}\right)^{n}}$$

Example

Constant Population

Design a waste stabilization pond to treat 10,000 m^3/day of a wastewater which has a BOD of 150 mg/L and 1 x 10⁸ FC per 100 mL. The effluent should contain no more than 5000 FC per 100 mL and 20 mg/L BOD . The design temperature is 28°C.

Solution

(a) Facultative Ponds

Design loading

$$= 350 (1.107 - 0.002T)^{T-25}$$
 (Eq. 9.2)

$$\lambda_s = 350[1.107 - (0.002 \times 28)]^{28-25}$$

= 406 kg/ha.day

$$A_{f} = 10 L_{i}Q / \lambda_{s}$$
(Eq. 9.1)
= 10 x 150 x 10,000/406
= 36,946 m²

$$t_f = A_f D/Q$$

Taking a depth of 1.5 m, this becomes:

$$t_f = 36,946 \times 1.5/10,000$$

= 5.5 d

(b) Maturation Ponds

Faecal Coliform Removal

$$k_T = 2.6 (1.19)^{T-20}$$

 $k_{28} = 2.6(1.19)^8 = 10.46 d^{-1}$

(Eq. 9.6)

$$N_{e} = \frac{N_{i}}{(1 + K_{T}t_{f})(1 + K_{T}t_{m})^{n}}$$
(Eq. 9.5)

Taking $t_m = 3$ days, this becomes:

$$N_e = \frac{10^8}{\left(1 + 10.46 \times 5.5\right)\left(1 + 10.46 \times 3\right)^n}$$

For n = 1, N_e = 52,765 > 5000 FC/100 mL For n = 2, N_e = 1,630 < 5000 FC/100 mL \rightarrow OK For a depth of 1.5 m, the area of the maturation pond is

$$A_{m} = Q t_{m} / D$$

= 10,000 x 3/1.5
= 20,000 m²

For one maturation pond, Area = $20,000/2 = 10,000 \text{ m}^2$

BOD Removal

Facultative and Maturation Pond:

$$K_{1(28)} = 0.3x1.05^{28-20} = 0.44 / day$$

$$l_e = \frac{l_i}{(1 + K_1 t_f)(1 + K_1 t_m)^n}$$

$$l_e = \frac{150}{(1 + 0.44 \times 5.5)(1 + 0.44 \times 3)^2} = 8.15 \text{ mg/L}$$